

Rhodium(III)−N-Heterocyclic Carbene-Driven Cascade C−H Activation Catalysis

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S Supporting Information

[AB](#page-3-0)STRACT: [Catalytic arom](#page-3-0)atic C−H bond activation and functionalization to useful molecules is highly important in chemical synthesis. This study introduces a novel use of the highly popular N-heterocyclic carbenes (NHC), for the firsttime, in rhodium(III)-catalyzed cascade double aromatic C−H activation−annulation, within the backbone of readily available imidazolium substrates to synthesize a variety of nicely decorated polycyclic heteroaromatic molecules containing benzo[ij]imidazo[2,1,5-de]quinolizinium architectures which might be useful for developing new luminescent materials.

KEYWORDS: C−H activation, rhodium(III), N-heterocyclic carbene, alkyne, annulation

The transition-metal-catalyzed aromatic C[−]H bond activation/functionalization protocol represents an important template in the chemistry of synthesizing various conjugated polyheteroaromatic molecules, useful as organic semiconductors and luminescent materials.¹ Direct, stepeconomic functionalization with readily available starting materials, avoiding complicated multistep tr[a](#page-3-0)ditional procedures, with a great degree of efficiency and selectivity are the main advantages that led to such a huge success of the C−H activation strategy. To add more novelty and practicality, very recently, there has been special attention toward double/ multiple C−H activation approach, which enables construction of highly conjugated beautiful architectures in one-pot via sequential reactions.² However, the discovery of such methodologies has always been challenging, and therefore, there are only a few reports o[f](#page-3-0) double/multiple C−H activation followed by annulative functionalization with alkynes described by the groups of Cheng and Chuang,^{2a} Miura and Satoh,^{2b−e} Huang,^{2f} Wang^2 ^{2g} Chen,^{2h} Li,^{2i,j} and Shi^{2k} during 2008−2014 (Scheme 1a). Strategically, a Lewis base [N](#page-3-0)- or O-donor fun[ction](#page-3-0)al grou[p,](#page-3-0) in-buil[t w](#page-3-0)ithin [th](#page-3-0)e s[ubs](#page-3-0)trate, is [us](#page-4-0)ed as a directing group (DG) [to](#page-1-0) facilitate such multiple C−H functionalization reactions. In contrast, the popular N-heterocyclic carbene (NHC) ligands have never been used as directing groups in multiple C−H functionalization chemistry mainly due to the robust, stable, and poorly reactive nature of the $M-C_{NHC}$ bonds.³ Herein, we demonstrate a novel strategy which not only uses the versatile M−C_{NHC} bond formation but also overcomes the [ch](#page-4-0)allenges of the so-formed robust $M-C_{NHC}$ bond functionalization, concomitant with a consecutive, double aromatic C−H activation−annulation, within the backbone of readily available

imidazolium substrates to synthesize a variety of nicely decorated polycyclic heteroaromatic molecules containing benzo[ij]imidazo[2,1,5-de]quinolizinium scaffolds (Scheme 1b). The chemistry of the above cascade annulation relies on the unprecedented utilization of normal as well as abnormal [N](#page-1-0)HC motifs as both directing and functionalizable groups in a sequential manner on a single molecular platform under rhodium(III) oxidative catalysis.⁴

Under standardized catalytic conditions, the reaction of 1 methyl-3-phenylimidazolium io[d](#page-4-0)ide (1a) (0.1 mmol) with diphenylacetylene (2a) (0.25 mmol) in the presence of $[Cp*Rh^{III}Cl₂]$ ₂ (0.005 mmol) as catalyst precursor, anhydrous NaOAc (0.8 mmol) as base, and AgOTf (0.6 mmol) as oxidant in dichloroethane at 100 °C for 48 h afforded the bis-annulated product, 1-methyl-3,4,8,9-tetraphenylbenzo $[i]$ imidazo $[2,1,5$ de]quinolizinium triflate, 4a, in excellent yield (91%) (Table 1), along with a trace amount of the monoannulated product, 3-methyl-4,5-diphenylimidazo[1,2-a]quinolinium triflate, 3a [\(5](#page-1-0)%) (for detailed optimization studies, see Supporting Information, Table S1). The molecular structure of $4a$ was unambiguously confirmed by single-crystal X-ray diff[raction](http://pubs.acs.org/doi/suppl/10.1021/acscatal.5b00243/suppl_file/cs5b00243_si_001.pdf) [analysis, whi](http://pubs.acs.org/doi/suppl/10.1021/acscatal.5b00243/suppl_file/cs5b00243_si_001.pdf)ch was well supported by ${}^1\mathrm{H}$ and ${}^{13}\mathrm{C}\{^1\mathrm{H}\}$ NMR spectra and mass spectrometry data (Table 1 and Supporting Information). It is noteworthy to mention that dioxane was also an alternative solvent for the above reactio[n](#page-1-0) to p[rovide 90%](http://pubs.acs.org/doi/suppl/10.1021/acscatal.5b00243/suppl_file/cs5b00243_si_001.pdf) [yield of](http://pubs.acs.org/doi/suppl/10.1021/acscatal.5b00243/suppl_file/cs5b00243_si_001.pdf) 4a. Other solvents (MeCN, MeOH) were less effective in the bis-annulation reaction. Also, other oxidants $(Cu(OAc))$,

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Scheme 1. (a) Directing-Group-Assisted Multiple C−H Functionalization, (b) Normal and Abnormal NHC as Tandem Directing as well as Functionalizable Groups

 O_2) or other catalyst precursor $([Cp*Ir^{III}Cl₂]₂)$ were found ineffective. Similarly, shorter reaction time or lower catalyst/ oxidant loading resulted in lower yield of 4a.

With the optimized reaction conditions, this unique double C−H activation/annulative functionalization of NHC backbones was further explored by reacting various substituted imidazolium salts with several internal alkynes to provide double annulated products in moderate to good yields (Table 1). All the products were isolated and purified by column chromatography (silica gel, $CHCl₃/a$ cetone solvent mixture as eluent; see Supporting Information). Replacement of the methyl by an n -butyl group as one of the N-substituent in the NHC [moiety reduced the yield](http://pubs.acs.org/doi/suppl/10.1021/acscatal.5b00243/suppl_file/cs5b00243_si_001.pdf) of the corresponding product (4b, 57%). Interestingly, electronic modification of the N-phenyl wingtip on the para position did not affect the reaction, as both $NO₂$ and OMe substituents afforded good yields of 4c (79%) and 4d (72%), respectively. A similar trend was observed with dialkyl acetylene where the reaction was found to be high-yielding providing 95% of 4e and 62% 4f. Here also the electron-withdrawing and -donating groups on an N-phenyl wingtip resulted in similar good yields of the corresponding products 4g and 4h. Interestingly, phenyl− alkyl unsymmetrical alkynes provided single regioisomers of 4i, 4j, and 4k in high yields. The structure of 4k was unambiguously confirmed by single crystal X-ray diffraction study (Table 1 and Supporting Information). On the other hand, alkynes bearing two different aromatic rings, produced four regioisomeric products of 4l in 70% combined yields.

To s[he](http://pubs.acs.org/doi/suppl/10.1021/acscatal.5b00243/suppl_file/cs5b00243_si_001.pdf)d light on the [mechanistic](http://pubs.acs.org/doi/suppl/10.1021/acscatal.5b00243/suppl_file/cs5b00243_si_001.pdf) [aspects](http://pubs.acs.org/doi/suppl/10.1021/acscatal.5b00243/suppl_file/cs5b00243_si_001.pdf) [of](http://pubs.acs.org/doi/suppl/10.1021/acscatal.5b00243/suppl_file/cs5b00243_si_001.pdf) this novel Rh^{III} catalyzed normal- and abnormal-NHC-directed double C−H

Table 1. Substrate Scope^a

 a Reaction conditions: 1 (0.1 mmol), 2 (0.25 mmol), catalyst (0.005 mmol), additives (as mentioned), ClCH₂CH₂Cl (3.0 mL), dinitrogen atmosphere, 100 °C, 48 h. The yields indicate isolated yields after purification of the products by column chromatography. \overline{b} A mixture of four regioisomers of 4l was obtained in 70% combined yields.

activation/annulation reaction, a series of control studies have been performed. At first, an H/D exchange study was conducted to examine the nature of the C−H bond activation in the reaction. The results shown in Scheme 2a suggested that the $C(4)_{im}$ −H and C_{phenyl} −H activations are reversible in nature. While the activation of these $C(4)_{im}$ −[H](#page-2-0) and C_{phenyl} −H bonds are obvious for an anticipated cyclometalation as required for the corresponding annulation reaction with the alkyne, the breaking of the $C(5)_{im}$ −H bond was interesting, indicating the possibility of the substrate 3e to act as a monodentate $C(5)_{\text{carbon}}$ ligand as well, in addition to the bidentate $C(4)_{\text{carbone}}^{\text{2}}C_{\text{phenyl}}$ cyclometalating ligand.⁵ To probe the above argument and to detect any plausible intermediate, a stoichiometric reaction of 2b, 3e and $[RhCp*Cl_2]_2$ $[RhCp*Cl_2]_2$ $[RhCp*Cl_2]_2$ was subjected to ESI-MS analysis (Scheme 2b). The spectrum displayed the product (4e) peak at $m/z = 375.2833$ as expected; however, interestingly, the peak [a](#page-2-0)t $m/z = 503.1966$ indicates the cation $[\text{RhCp*}(C(4))_{\text{carbon}}^{\text{max}}C_{\text{phenyl}})]^+$ and a trace peak at $m/z = 769.3744$ suggested the generation of a species, $[\text{RhCp*}(C(4))_{\text{carbon}}^{\text{-}}C_{\text{phenyl}})(L)]^+$ (L = 3e) in the reaction mixture in which the compound 3e was bound to Rh^{III} in expected $C(4)_{\text{carbon}}$ ²C_{phenyl} cyclometalating fashion with an additional monodentate $\dot{C}(5)_{\text{carbon}}$ -bound ligand.

The above encouraging results prompted us to attempt to isolate such a rhodium intermediate complex to confirm its existence in the catalysis. After several trials, a reaction between

Scheme 2. Mechanistic Studies: (a) H/D Exchange, (b) ESI-MS Study

3a and $[RhCp*Cl_2]_2$ in the presence of NaOAc in dichloroethane provided the complex 5 (Scheme 3a; see Supporting Information for detailed experimental procedure). The ¹H and Information for detailed experimental procedure). The ¹H and
¹³C{¹H} NMR and ESI-MS data of **5** suggested its [structure as](http://pubs.acs.org/doi/suppl/10.1021/acscatal.5b00243/suppl_file/cs5b00243_si_001.pdf) [the one show](http://pubs.acs.org/doi/suppl/10.1021/acscatal.5b00243/suppl_file/cs5b00243_si_001.pdf)n in Scheme 3a. Gratifyingly, the confirmation of such a structure came from the single crystal X-ray diffraction studies (Scheme 3e). As argued earlier, the molecular structure of 5 consisted of a $C(4)_{\text{carbene}}$ ² C_{phenyl} cyclometalating 3a chelated to Rh^{III} with an additional $C(5)$ _{carbene} bound 3a at the metal center. It seems that under the basic condition applied in the reaction, the $C(5)_{im}$ −H bond activated and acted as an abnormal carbene ligand to replace any anionic ligand (Cl or OAc) or solvent molecule bound to the rhodium center.

The plausible involvement of the complex 5 during the catalytic annulation reaction was examined by performing a catalytic as well as stoichiometric reaction, as shown in Scheme 3b and 3c, respectively. Thus, the stoichiometric reaction furnished the desired product 4a in 44% yield, while the reaction with a 5 mol % loading of 5 as catalyst provided 4a in 41% yield. These observations clearly suggested that 5 is a tenable intermediate in the present annulation reaction.

The significance of the $C(5)_{im}$ −H activation (as indicated earlier by H/D exchange, ESI-MS, and crystallographic studies) during the catalysis was realized by conducting an annulation reaction with the substrate 1-benzyl-3-phenyl imidazolium bromide, 1e, with alkyne 2a (Scheme 3d). This reaction provided the expected bis-annulated product 4m with an additional minor product 4m′ (in 68% overall yield). Interestingly, 4m′ consisted of a seven-membered cyclic ring system where the $C(S)_{im}$ −H bond of 1e was found to be activated and annulated with the pendant benzyl group via an aromatic C−H activation of the benzyl moiety. The molecular structure of such an interesting annulated product was confirmed by single crystal X-ray diffraction analysis (Scheme 3e). Finally, a kinetic deuterium isotope effect experiment was conducted as shown in Scheme 3f. A small k_H/k_D value of 1.35 Scheme 3. Mechanistic Studies: (a) Synthesis and (b) Catalytic/ (c) Stoichiometric Reaction of Intermediate 5, (d) C(5)−H Activation/Annulation, (e) X-ray Structures, (f) KIE Experiment

± 0.32 suggested that the C_{phenyl}−H activation step might not be involved in the rate-limiting step.

On the basis of the above mechanistic investigation, a plausible catalytic cycle is shown in Scheme $4.^{2,4,6}$ The initial

Scheme 4. Plausible Catalytic Cycle

step involves the formation of a five-membered rhodacycle I from the imidazolium substrate 1, via ortho C−H activation directed by the coordinated NHC ligand, in the presence of NaOAc.^{3d'} Halide abstraction from I by AgOTf triggers coordination of alkyne 2 to generate intermediate II. Insertion of the c[oor](#page-4-0)dinated alkyne into the Rh−C_{aryl} bond of II provides the seven-membered rhodacycle intermediate III. Subsequent reductive elimination from III in the presence of $Ag¹$ affords the monoannulated product 3 and a $\mathrm{Cp}^*\mathrm{Rh}^{\mathrm{III}}$ species. In the presence of NaOAc, 3 is activated to coordinate with the rhodium center again as an abnormal NHC which directs the ortho C−H activation of 3 to form the intermediate 5. A second round of coordinative insertion of alkyne 2 (IV to V) followed by reductive elimination in the presence of Ag^I results in the product 4 and regenerates the C_P^+ Rh^{III} species to continue the catalytic cycle via reaction with 1 and NaOAc. Isolation and characterization of the other intermediates and DFT studies are underway to fully validate the proposed pathway, including the $NHC/alkyne insertion possibility (in II to III and IV to V) as$ suggested by Bergman and Ellman, and Cavell.⁷

Some of these benzo[ij]imidazo[2,1,5-de]quinolizinium motif-containing bis-annulated products ex[h](#page-4-0)ibited strong fluorescence emission at 455−465 nm range, which suggests the potential of this class of ionic compounds to be useful, solution-processable, new emitting materials (Figure S18). Detailed photophysical studies toward this direction are ongoing.

In conclusion, we have successfully demonst[rated,](http://pubs.acs.org/doi/suppl/10.1021/acscatal.5b00243/suppl_file/cs5b00243_si_001.pdf) [for](http://pubs.acs.org/doi/suppl/10.1021/acscatal.5b00243/suppl_file/cs5b00243_si_001.pdf) [th](http://pubs.acs.org/doi/suppl/10.1021/acscatal.5b00243/suppl_file/cs5b00243_si_001.pdf)e first time, that a normal NHC and an abnormal NHC ligand not only directed two consecutive C−H activation reactions but also guided the sequential annulations via Rh- $C_{\rm NHC}/{\rm Rh}-C_{\rm aryl}$ bond functionalization with internal alkynes to construct a variety of nicely decorated benzo[ij]imidazo[2,1,5-de]quinolizinium scaffold-containing polycyclic heteroaromatic molecules. A series of control studies suggested the proposed mechanistic sequence involving NHC-directed C−H activation/insertion/annulative reductive elimination/oxidative catalyst regeneration in the catalytic cycle. This new and simple method appears to have the potential for the synthesis of new class of organic functional materials. Full mechanistic investigation and diverse application of this newly developed protocol is the subject of current research in our laboratory.

ASSOCIATED CONTENT

6 Supporting Information

The following files are available free of charge on the ACS Publications website at DOI: 10.1021/acscatal.5b00243.

Experimental details; spectra (PDF) CIF file of $4a$ (CIF) CIF file of $4k$ (CIF) CIF file of $4m$ [\(CIF](http://pubs.acs.org/doi/suppl/10.1021/acscatal.5b00243/suppl_file/cs5b00243_si_002.cif)) CIF file of 5 ([CIF\)](http://pubs.acs.org/doi/suppl/10.1021/acscatal.5b00243/suppl_file/cs5b00243_si_003.cif)

■ AUTHOR INF[ORM](http://pubs.acs.org/doi/suppl/10.1021/acscatal.5b00243/suppl_file/cs5b00243_si_005.cif)[A](http://pubs.acs.org/doi/suppl/10.1021/acscatal.5b00243/suppl_file/cs5b00243_si_004.cif)TION

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Notes

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(5) Under identical reaction conditions but in the absence of Rhcatalyst, the D-incorporation into the imidazolium backbone protons of 1a, $C(2)_{\text{im}}$ –H, $C(4)_{\text{im}}$ –H, and $C(5)_{\text{im}}$ –H was observed to be 16%, 31%, and 8% respectively. However, as expected, there was no H/D exchange into the phenyl ortho-protons. Similarly, in case of 3e, the $C(4)_{\text{im}}$ –H and $C(5)_{\text{im}}$ –H protons were exchanged with D in 31% and 12% respectively, without any H/D exchange into the phenyl ortho proton. See Supporting Information for details.

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